

## Unit 1 Measurements

Things to know: scientific method, qualitative/quantitative data, SI measurements (m, L, g), ways to find volume, converting with prefixes, density, accuracy vs. precision, % error, significant figures, scientific notation, atom/element/compound, chemical/physical properties, chemical/physical changes, classification of matter (mixture/pure substance and homogeneous/heterogeneous), periodic table (periods/groups/families, metals/nonmetals/metalloids)

\*Items in *italics* is for honors only

### Objective

2.2.2—Analyze the evidence of chemical change.

### Problems

Highest Mastery Achieved for 2.2.2 \_\_\_\_\_  
(80% considered mastered)

- List the four indicators of a chemical change:
  - Formation of a precipitate
  - Formation of a gas
  - Change in color
  - Change in energy
- Define precipitate. How can you determine if something is a precipitate or not?  
New solid formed during a reaction...is always on the product side (right side)...can determine a precipitate by using the solubility rules (insoluble = precipitate)
- What would happen if a burning splint was exposed to:
  - Oxygen the flame would be brighter and bigger
  - Hydrogen the flame would make a popping sound...sometimes called barking
  - Carbon dioxide the flame would go out
- What would happen if carbon dioxide was bubbled through lime water?  
The water would look cloudy
- What piece of safety equipment must be used in all laboratory experiments?  
Goggles
- When using a Bunsen burner what precautions should be made?  
Long hair and clothes should be pulled back and there should be no flammable materials near by
- Exothermic reaction releases energy,  $\Delta H$  is *neg* and energy is shown on the *product (right)* side.
- Endothermic reaction absorbs energy,  $\Delta H$  is *+* and energy is shown on the *reactant (left)* side.
- Label each of the following as a physical or chemical property and give a brief reason
  - Flammability  
CP, cannot get substance back/change in color
  - Density  
PP, description...doesn't change substance
  - Can neutralize a base  
CP, cannot get substance back
  - Boiling Point  
PP, phase change...still same substance
- Label each of the following as a physical or chemical change
  - A tire is inflated with air.  
PC, doesn't change substance
  - Food is digested in the stomach.  
CP, can't get substance back, gas formed
  - Water is added to red soln, it turns pink.  
PP, diluted...same substance...no color change
  - Water is heated and changed into steam.  
PP, phase change...still same substance

# Msl Ss Review 8 Answers

**Jicheng Xie**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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