Solutions: Metathesis Reactions and Net Ionic Equations

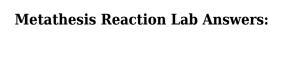
9 Pre-lab Questions

Before beginning this experiment in the laboratory, you should be able to answer the following questions.

- Write molecular, complete ionic, and net ionic equations for the reactions that occur, if any, when solutions
 of the following substances are mixed:
 - (a) nitric acid and potassium carbonate
 - (b) sodium bromide and lead nitrate
 - (c) acetic acid and calcium hydroxide
 - (d) calcium nitrate and sodium sulfate
 - (e) ammonium chloride and lithium hydroxide
- Which of the following are not water-soluble: Ba(NO₃)₂, FeCl₃, CuCO₃, CuSO₄, ZnS, and ZnSO₄?
- 3. Write equations for the decomposition of H2CO3 and H2SO3.
- 4. At what temperature (from your graph) do KNO3 and NaCl have the same molar solubility?

Metathesis Reaction Lab Answers

Jianjun Gao



Uncover the mysteries within Explore with is enigmatic creation, Embark on a Mystery with **Metathesis Reaction Lab Answers**. This downloadable ebook, shrouded in suspense, is available in a PDF format (Download in PDF: *). Dive into a world of uncertainty and anticipation. Download now to unravel the secrets hidden within the pages.

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