| C | Question |      | Answer  | Marks | Guidance  |
|---|----------|------|---|-------|---|
| 1 | (a)      |      | (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound ✓ from its gaseous ions (under standard conditions) ✓ | 2     | IGNORE 'energy needed' OR 'energy required'  ALLOW as alternative for compound: lattice, crystal, substance, solid  Note: 1st mark requires 1 mole 2nd mark requires gaseous ions IF candidate response has '1 mole of gaseous ions', award 2nd mark but NOT 1st mark |
|   | (b)      | (0)  | Ca <sup>2*</sup> (g) + O <sup>2-</sup> (g)   Step G   | 2     | Correct species AND state symbols required for both marks  2e* required for left-hand response ALLOW e for e*  Mark each marking point independently  |
|   |          | (ii) | (enthalpy change of) formation (of calcium oxide) ✓  (enthalpy change of) atomisation of oxygen ✓  Second electron affinity (of oxygen) ✓     | 3     | calcium oxide not required for this mark DO NOT ALLOW 'lattice formation' (confusion with LE) atomisation AND oxygen/Oy/Y/Oy/O both required (atomisation of calcium is also in cycle)  IGNORE oxygen or oxygen species, e.g. O' DO NOT ALLOW calcium                 |

# June 13 Mark Scheme F325

**KJ Lindholm-Leary** 

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