চূড়ান্ত সাজেশন HSC 2023(Chemistry 1st Paper CQ)

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Mb: 01747823874

1st Paper, Chapter-2 : গুণগত রসায়ন(2 সেট cq আসবে)

Part-1: পরমান, কোয়ান্টাম সংখ্যা, H এর পারমাণবিক বর্ণালি (1 সেট CQ থাকবে)

- ★1) त्रामात्रकार्फ भत्रभान् भएजन ७ श्रीभावक्त्रजा
- ★★2) বোর পরমানু মডেল ও সীমাবদ্ধতা ও math: mvr= nh/2π
- ★★★3) পরমানুর ৩য় অথবা ৪র্থ কক্ষপথে কতগুলো অরবিটাল বা ইলেকট্রন থাকতে পারে কোয়ান্টাম সংখ্যার সাহায্যে বের করো(ছক)
- ★ ★ ★ 4) কক্ষপথের ইলেকট্রনের ধাপান্তরেরে ফলে সৃষ্ট বর্ণালির তরঙ্গ দৈর্ঘ্য অথবা বর্ণ অথবা শক্তি নির্ণয় করো।
- ★★5)সর্বশেষ ইলেক্ট্রনের জন্য পাউলির বর্জন নীতি দেখাঙ

Part-2: দ্রাব্যতা ও আয়ন শনাক্তকরণ(1 সেট CQ আসবে)

- ★★★1) দ্রাবতা দেওয়া ঘাকবে Ksp বের করো
- ★★★2) Ksp দেওয়া দ্রাবতা অথবা আয়নের ঘনমাত্রা বের করো
- ★★★3) দুটি দ্রবনকে মেশালে অধ্ঃক্ষেপ পড়বে কিনা গাণিতিক ভাবে বিশ্লেষন করো।
- ★ ★4) আয়ন শনাক্তকরণ: ক্ষারীয়: NH4*,Al³+,Ca²+,Na*,Cu²+ (less important Fe²+Fe³+) অসীয়: Cl-, SO₂²-, CO₂²-

Extra: * *5) দ্রবন ঠাণ্ডা করলে কতটক অধঃক্ষেপ পড়বে

- ★★6)তাপমাত্রা বাড়ালে অতিরিক্ত কতিটুকু লবন দ্রবীভূত করলে দ্রবনটি সম্পৃক্ত হবে
- ★★7) সম আয়ন math

NB: Part-2 থেকে 1, 3 প্রশ্নটা আসার সম্ভাবনা বেশি।

1st Paper, Chapter-3 : মৌলের পর্যায়বৃত্ত ধর্ম ও রাসায়নিক বন্ধন(2 সেট CQ আসবে)

- ★★★1) আয়নীকরণ শক্তি:
 - ★★★i)Be থেকে B এর আয়নীকরণ শক্তি কম কেন?
 - ★★★ii)N থেকে O এর আয়নীকরণ শক্তি কম কেন?
- * * * 2)ইলেক্ট্রন আসন্তি:
 - ★★★i) F থেকে CI এর ইলেইন আসক্তি বেশি কেন?
- ★★★3)সংকরন: প্রশ্ন হবে সংকরণ ব্যাঘ্যা কর, গঠন, আকৃতি, বন্ধুন কোন এত কেন অথবা তুলনা করো।
 - i) যাদের Ip ইলেকট্রন নেই: CH4PCIs.SF6.8F3 ★★★CH2=CH2. ★★★ CHⅢCH or C2H2
 - ii)যাদের lp ইলেক্ট্রন আছে: ★★★NH3. ★★★ H2O.PH3.H2S

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