

REPORT SHEET

Heat of Neutralization

EXPERIMENT
28

A. Heat Capacity of Calorimeter

1. Temp. of calorimeter and water before mixing
2. Temp. of warm water
3. Maximum temp. determined from your curve
4. Heat lost by warm water (temp decrease \times 50.0 g \times 4.184 J/K-g) =

$$\begin{array}{r} 26 \\ 39 \\ 41 \end{array} \quad \begin{array}{l} ^\circ\text{C} \\ ^\circ\text{C} \\ ^\circ\text{C} \end{array}$$

$$57.822 \text{ J}$$

5. Heat gained by cooler water (temp. increase \times 50.0 g \times 4.184 J/K-g) =

$$58.861 \text{ J}$$

6. Heat gained by the calorimeter [(4) - (5)] =

$$-1.047 \text{ J}$$

7. Heat capacity of calorimeter:

$$\frac{\text{heat gained by the calorimeter}}{\text{temperature increase}}$$

$$-3.72 \text{ J/K}$$

B. Heat of Neutralization of HCl-NaOH

1. Temp. of calorimeter and NaOH
Temp. of HCl
2. ΔT determined from your curve after adding HCl to the NaOH

$$\begin{array}{r} 25 \\ 25 \end{array} \quad \begin{array}{l} ^\circ\text{C} \\ ^\circ\text{C} \end{array}$$

$$8^\circ\text{C}$$

3. Heat gained by solution (temperature increase \times 100 g \times 4.184 J/K-g) =

$$1175.70 \text{ J}$$

4. Heat gained by calorimeter (temperature increase \times heat capacity of calorimeter) =

$$1045.32 \text{ J}$$

5. Total joules released by reaction [(3) + (4)] =

$$116,525 \text{ J}$$

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AW Chickering



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Laboratory Experiments for Brown and LeMay, Chemistry, the Central Science Nelson, John Henry Nelson, Kenneth C. Kemp, Kemp, 1981 **Experiments in Chemistry** Frank R. Milio, Clyde R. Metz, W. G. Nordulf, 1991-03 Preliminary Report on the Thermodynamic Properties of Selected Light-element and Some Related Compounds, 1962 Laboratory Experiments for Chemistry, the Central Science, 5th Ed John Henry Nelson, 1991 **Chemistry** John H. Nelson, 1997 A standard in the industry this best selling lab manual was written in conjunction with Brown LeMay Bursten's Chemistry The Central Science 7 e but can be used as a stand alone lab manual This edition has been updated to reflect environmental concerns **Scientific and Technical Aerospace Reports**, 1995 **General Chemistry Laboratory Operations** Lawrence E. Conroy, Russell Stuart Tobias, Robert Crocker Brasted, 1971 Analytical Chemistry Division Annual Progress Report for Period Ending ... Oak Ridge National Laboratory. Analytical Chemistry Division, Public Health Reports, 1920 Reprint from the Public Health Reports United States. Public Health Service, 1928 **Journal of Research of the National Bureau of Standards** United States. National Bureau of Standards, 1973 **Health Services Reports**, 1920 American Chemical Journal Ira Remsen, Charles August Rouiller, 1907 Cancer Chemotherapy Reports, 1963 Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Hearings, Reports and Prints of the House Committee on Post Office and Civil Service United States. Congress. House. Committee on Post Office and Civil Service, 1965 **U.S. Government Research & Development Reports**, 1969-10 U.S. Government Research Reports, 1962-07 **Fundamentals of Chemistry in the Laboratory** Ralph A. Burns, 1999 **Official Documents, Comprising the Department and Other Reports Made to the Governor, Senate and House of Representatives of Pennsylvania**, 1906

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