

Q6.1 (2014)

Sum of the exterior angles of any polygon is equal to _____

– requires recall of facts.

For many learners will not recall this fact, more difficult than the sum of interior angles of a triangle or quadrilateral.

Q6.2 (2013)

Find x :

$$(x - 2)^2 = (2x - 1)(x - 3)$$

– requires 4 well-rehearsed procedures.

– combination of several procedures, executing them matters.

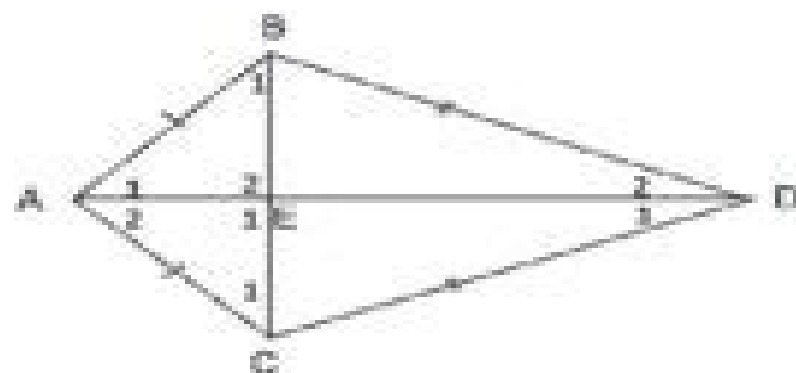
Q6.3 (2013)

Find x : $x^3 = 64$

– finding – not in Grade 9 curriculum but in Grade 10/11.

– trial and improvement with whole numbers will quickly reveal the answer.

Q6.3 (2012)



In the above figure $AB = AC$ and $BD = CD$

6.3.1 Prove that $\triangle ABD \equiv \triangle ACD$

6.3.2 Prove that $\triangle ABE \equiv \triangle ACE$

6.3.3 Prove that $\angle E_1 = \angle E_2$

6.3.1 Routine, easy – all information given.

6.3.2 Routine, moderate – need to deduce equal sides.

6.3.3 Routine, difficult – need to deduce from congruence and adjacent angles.

Grade 9 Ana Exam 2013 Maths

Terry C. Jones



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \frac{7}{2} \text{O}_2(\text{g}) + 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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changes of its individual steps hess s law example 12.1 determine Δh for the target reaction $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$
given the following information

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CO_2 g 2 H_2O l h 1411 kJ C_2H_6 g $3\frac{1}{2}$ O_2 g 2 CO_2 g 3 H_2O l h 1560 kJ

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web thus for reaction 9 2 in which 1 mole of H_3O^+ and 1 mole of OH^- combine to form 2 moles of H_2O h 55.8 kJ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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