

8. Given the following data for a fluid, what is its state at 40°C and 3 kPa?

Saturated property table

$T_s, ^\circ\text{C}$	P_s, kPa	$v_{f,s}, \text{m}^3/\text{kg}$	$v_{g,s}, \text{m}^3/\text{kg}$	$h_{f,s}, \text{kJ/kg}$	$h_{g,s}, \text{kJ/kg}$	$s_{f,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$	$s_{g,s}, \text{kJ/(kg} \cdot ^\circ\text{K)}$
40	7.38	.001008	19.52	167.57	2574.3	.5725	8.257
80	47.39	.001029	3.407	334.9	2643.7	1.1343	7.5645
120	198.5	.001060	.8919	503.71	2706.3	1.5276	7.1296

- (A) Saturated liquid
- (B) Superheated vapor
- (C) Compressed liquid
- (D) Saturated vapor

9. Using the same data of Problem 8, what is the fluid's entropy in kJ/(kgK) at 120°C and 80% quality?

- (A) 1.53
- (B) 6.009
- (C) 7.13
- (D) 28.8

10. A quantity of 55 m³ of water passes through a heat exchanger and absorbs 2,800,000 kJ. The exit temperature is 95°C. The entrance water temperature in °C is nearest to:

- (A) 49
- (B) 56
- (C) 68
- (D) 83

11. In any non-quasistatic thermodynamic process, the overall entropy of an isolated system will:

- (A) Increase and then decrease
- (B) Decrease and then increase
- (C) Stay the same
- (D) Increase only

12. The temperature-entropy diagram in the figure represents a:

- (A) Rankine cycle with superheated vapor
- (B) Carnot cycle
- (C) Diesel cycle
- (D) Refrigeration cycle



13. Ethane is burned in 150% theoretical air. Assume complete combustion and that the process takes place in atmospheric air. The air/fuel ratio in 150% theoretical air is most nearly:

- (A) 32 kg air/kg fuel
- (B) 28 kg air/kg fuel
- (C) 24 kg air/kg fuel
- (D) 16 kg air/kg fuel

14. There are 3 kg of air in a rigid container at 250 kPa and 50°C. The gas constant for air is 0.287 kJ/(kgK). The volume of the container, in m³, is nearest to:

- (A) 2.2
- (B) 1.1
- (C) 2.8
- (D) 3.1

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Thomas Griffiths



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