

SB14UC UNIT 1

Lesson 1: Question 1

Mg 2+ has 10 electrons

Teacher Comments:

Mark: 1/1

Lesson 1: Question 2

The carbon-14 radioisotope will not be useful for dating bones that are over a million years old, because, with a half life of 5730 years it shows it can be used to date bones of thousands of years, and after 40,000 years less than 1% of the Carbon-14 is left in the tissue and is not useful in dating very old material such as dating bones that are over one million years old.

Teacher Comments:

CORRECT

Mark: 2/2

Lesson 1: Question 3

The four major types of biochemical reaction studied in this lesson are hydrolysis, condensation, oxidation-reduction (Redox) and neutralization.

a). Hydrolysis is a process by which water is used to help breakdown molecules into smaller parts. During this reaction water is very important to carry out the process. An example is when food is consumed, water and enzyme is required to breakdown the food into smaller parts in our body, and this process is called hydrolysis. Here is a simple hydrolysis reaction; protein + water? Amino acid + amino acid.

b). Condensation takes place when two molecules join together to form one molecule. During this process water is removed, when two smaller molecules combine together to produce a larger molecule. This kind of reaction happens in our body when building muscle tissue. A simple condensation reaction is; amino acid + amino acid? Protein + water

c). Oxidation-Reduction (Redox): Oxidation is the process of losing electron from one or more reactant to another while Reduction is the process of gaining electrons. Where there is an electron transfer between two substances, there is always an oxidation and reduction that takes place, this is also known as a "redox reaction" Example of this is the cellular respiration i.e. $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$ +

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E Durkheim



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