

ICSE Class 10 Chemistry Question Paper Solution 2016

CHEMISTRY SCIENCE Paper – 2

Question 1

- (a) Fill in the blanks with the choices given in brackets. [5]
- (i) Metals are good _____ (*oxidizing agents / reducing agents*) because they are electron _____ (*acceptors/ donors*).
 - (ii) Electrovalent compounds have _____ (*high / low*) melting points.
 - (iii) Higher the pH value of a solution, the more _____ (*acidic / alkaline*) it is.
 - (iv) _____ (*AgCl / PbCl₂*), a white precipitate is soluble in excess NH_4OH .
 - (v) Conversion of ethene to ethane is an example of _____ (*hydration / hydrogenation*).
- (b) Choose the *correct answer* from the options given below: [5]
- (i) An element with the atomic number 19 will most likely combine chemically with the element whose atomic number is:
 - A. 17
 - B. 11
 - C. 18
 - D. 20
 - (ii) The ratio between the number of molecules in 2g of hydrogen and 32g of oxygen is:
 - A. 1:2
 - B. 1:0.01
 - C. 1:1
 - D. 0.01:1 [Given that H=1, O=16]
 - (iii) The two main metals in *Bronze* are:
 - A. Copper and zinc
 - B. Copper and lead
 - C. Copper and nickel
 - D. Copper and tin
 - (iv) The particles present in *strong electrolytes* are:
 - A. only molecules
 - B. mainly ions

Grade 10 Chemistry Questions And Answers

K Morrison



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University English for Academic Purposes in China Xiaofei Rao,2018-07-07 This book uses an in depth phenomenological interview approach to explain the generational characteristics of today s Chinese university youths and the critical dispositions they believe indispensable in acquiring English as an academic language in and outside school settings By presenting the authentic voices of the recruited participants the book clarifies how English for academic purposes EAP as an emerging global phenomenon and a research informed practice enables and empowers them for conscious self transformation and critical awareness development through language study The book also explores issues arising in the fields of general English language teaching as well as traditional and critical EAP and discusses university English language learners learning needs and rights The book further promotes a dynamic and transformative University EAP pedagogy of particularity practicality and possibility moving from the oppression of language education to its liberation and the increasing critical consciousness among the present and future university youths in a time of great social changes **British Books** ,1898 *The Publishers Weekly* ,1917 **Developing Standards in Research on Science Education** Fischer Hans

E,2005-07-14 Ph D students from 14 European countries Israel Turkey and Venezuela in addition to supervisors and lecturers from 11 countries including Israel and USA worked for one week to understand each other with the goal of improving and sharpening features of their respective theoretical backgrounds research questions and design and methodological demands The projects presented reflect a multitude of topics and goals of research in science education in Europe as well as the variety and elaboration of theoretical frameworks used and a remarkable level of methodological expertise The following topics are included Teachers thinking and beliefs and teachers actions in the classroom the interaction between specific programs of science museums and teachers and classes plans for engagement with them teaching learning and understanding of new subject matter for science classes different interaction processes in the classroom discourse analysis decision making processes in science classes and the use of models in chemistry lessons and last but not least specific characters and the function of text when learning physics by means of computer based multimedia All science subjects including earth science are involved in these articles and the level of analysed e *SDEA Journal* South Dakota Education Association,1928 **Multiple Representations in Chemical Education** John K. Gilbert,David Treagust,2009-02-28 Chemistry seeks to provide qualitative and quantitative explanations for the observed behaviour of elements and their compounds Doing so involves making use of three types of representation the macro the empirical properties of substances the sub micro the natures of the entities giving rise to those properties and the symbolic the number of entities involved in any changes that take place Although understanding this triplet relationship is a key aspect of chemical education there is considerable evidence that students find great difficulty in achieving mastery of the ideas involved In bringing together the work of leading chemistry educators who are researching the triplet relationship at the secondary and university levels the book discusses the learning involved the problems that students encounter and successful approaches to teaching Based on the reported research the editors argue for a coherent model for understanding the triplet relationship in chemical education *Manual of Examinations for the ...* ,1915 **Report Upon the Survey of the University of Wisconsin** Wisconsin. State Board of Public Affairs,1914 Publishers' circular and booksellers' record ,1899

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Table of Contents Grade 10 Chemistry Questions And Answers

1. Understanding the eBook Grade 10 Chemistry Questions And Answers
 - The Rise of Digital Reading Grade 10 Chemistry Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 10 Chemistry Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 10 Chemistry Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 10 Chemistry Questions And Answers
 - Personalized Recommendations
 - Grade 10 Chemistry Questions And Answers User Reviews and Ratings
 - Grade 10 Chemistry Questions And Answers and Bestseller Lists
5. Accessing Grade 10 Chemistry Questions And Answers Free and Paid eBooks

- Grade 10 Chemistry Questions And Answers Public Domain eBooks
 - Grade 10 Chemistry Questions And Answers eBook Subscription Services
 - Grade 10 Chemistry Questions And Answers Budget-Friendly Options
6. Navigating Grade 10 Chemistry Questions And Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Grade 10 Chemistry Questions And Answers Compatibility with Devices
 - Grade 10 Chemistry Questions And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Grade 10 Chemistry Questions And Answers
 - Highlighting and Note-Taking Grade 10 Chemistry Questions And Answers
 - Interactive Elements Grade 10 Chemistry Questions And Answers
8. Staying Engaged with Grade 10 Chemistry Questions And Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 10 Chemistry Questions And Answers
9. Balancing eBooks and Physical Books Grade 10 Chemistry Questions And Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 10 Chemistry Questions And Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade 10 Chemistry Questions And Answers
- Setting Reading Goals Grade 10 Chemistry Questions And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade 10 Chemistry Questions And Answers
- Fact-Checking eBook Content of Grade 10 Chemistry Questions And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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