

DAT General Chemistry Equation Sheet

General Concepts

Density: ($d = m/v$)

Dilutions: $M_1V_1 = M_2V_2$ (or $C_1V_1 = C_2V_2$)

% Error Calculations

$$\% \text{ Error} = \frac{\text{Actual} - \text{theoretical}}{\text{theoretical}} \times 100$$

Spectrophotometer: $Abs = \epsilon cl$

Abs = Absorbance

ϵ = molar extinction coefficient (molar absorptivity)

l = path length

Atomic Structure

Absorption/Emission Line Spectra: $\Delta E = E_{\text{photon}}$

Energy of a photon: $E_{\text{photon}} = hf = \frac{hc}{\lambda}$

$$c = 3.0 \times 10^8 \text{ m/s}$$

$$h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$$

Photoelectric Effect: $KE_{\text{electron}} = E_{\text{photon}} - \phi$

ϕ = work function (minimum energy to ionize electron)

Chemical Kinetics and Equilibrium

General Rate Law: $\text{rate} = k[A]^m[B]^n$

Arrhenius Equation: $k = Ae^{-E_a/RT}$

E_a = activation energy

$R = 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K}$

Equilibrium Constant:

$$K_c = \frac{[\text{products}]}{[\text{reactants}]} \quad K_p = \frac{P_{\text{products}}}{P_{\text{reactants}}}$$

Solutions

Concentration

$$\text{Molarity (M)} = \frac{\text{moles}_{\text{solute}}}{L_{\text{solution}}}$$

$$\text{molality (m)} = \frac{\text{moles}_{\text{solute}}}{kg_{\text{solvent}}}$$

Henry's Law: $P_A = k_H[A]$

Colligative Properties

Freezing Point Depression: $\Delta T_F = -iK_F m$

Boiling Point Elevation: $\Delta T_B = iK_B m$

Raoult's Law ($*$ = pure): $P_A = X_A P_A^*$

Osmotic Pressure: $\pi = iMRT$

$$K_F/K_B = 1.86/0.51$$

depression/elevation constant

X_A = mole fraction of A

i = van't Hoff Factor

$R = 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K}$

M = molarity, m = molality

Gases/Gas Laws

Average Kinetic Energy:

$$KE_{\text{avg}} = \frac{3}{2} RT \text{ (per mole of gas)}$$

$$\text{Root-mean-square speed: } v = \sqrt{\frac{3RT}{M_m}}$$

Gas Laws

Ideal Gas Law: $PV = nRT$

Boyle's Law: $P \propto \frac{1}{V}$

Charles' Law: $V \propto T$

Avogadro's Law: $V \propto n$

$$\text{Combined Gas Law: } \frac{P_1 V_1}{n_1 T_1} = \frac{P_2 V_2}{n_2 T_2}$$

Dalton's Law of Partial Pressures:

$$P_{\text{Tot}} = P_A + P_B + \dots$$

$$P_A = X_A \cdot P_{\text{Tot}}$$

$$X_A = \text{mole fraction A}$$

Graham's Law of Effusion:

$$\frac{r_1}{r_2} = \sqrt{\frac{M_{m,2}}{M_{m,1}}}$$

$R = 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K}$

n = moles

M_m = molar mass

r = rate of effusion

a and b are constants specific to a given gas

Acid-Base

pH Scale

$$pH = -\log [H^+] \quad [H^+] = 10^{-pH}$$

$$pOH = -\log [OH^-] \quad [OH^-] = 10^{-pOH}$$

$$pH + pOH = 14 \quad [H^+][OH^-] = 1 \times 10^{-14}$$

Weak Acids:

$$K_a = \frac{[H_3O^+][A^-]}{[HA]} \quad [H^+] = \sqrt{K_a [HA]}$$

Weak Bases:

$$K_b = \frac{[OH^-][HA]}{[A^-]} \quad [OH^-] = \sqrt{K_b [B]}$$

$$K_a \cdot K_b = K_w$$

$$\text{Buffer: } pH = pK_a + \log \frac{[A^-]}{[HA]}$$

$$\text{Titration/Neutralization: } n_A M_A V_A = n_B M_B V_B$$

DAT Bootcamp

Formula Sheet For Gen Chem 1 Chm 2045

Yijin Wang



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of mice and men novel quiz exam proprofs - Apr 07 2022

web sep 15 2023 correct answer c john steinbeck explanation john steinbeck is the correct answer because he is the author of the novel of mice and men the book published in 1937 tells the story of two displaced ranch workers george and lennie during the great depression

of mice and men quizzes questions answers proprofs - Jun 21 2023

web mar 22 2023 this quiz will help test your knowledge and understanding of the characters in john steinbeck s novel of mice and men questions 10 attempts 1432 last updated mar 22 2023 sample question

of mice and men quiz bbc bitesize - Nov 14 2022

web lennie george question 2 of 7 who says guys like us that work on ranches are the loneliest guys in the world they got no family they don't belong no place with us it ain't

of mice and men chapter 1 quiz john steinbeck - Oct 13 2022

web jul 1 2020 this free package includes a fill in the blank quiz with answer key for chapter 1 of of mice and men by john steinbeck the quiz with answer key comes as both microsoft word and pdf documents this is a great open book reading quiz that assesses and reinforces comprehension and can be graded quickly

of mice and men test gotoquiz - May 08 2022

web this is a test on the book of mice and men by john stienbeck it the main events of the novel as well as some minor ones it is helpful to take if you are studying for a quiz or just want to see how well you know the book how well do you know the book of mice and men do you think you know everything about it

of mice and men characters test questions wjec bbc - Dec 15 2022

web the main characters in of mice and men are george and lennie they are best friends but complete opposites learn about the other characters on the ranch and how they all long for friendship