

ELECTROCHEMISTRY

A. Matching

Match each term in Column B to the correct description in Column A.

	Column A	Column B
H D j G i e H	1. a conductor in a circuit that carries electrons to or from a substance other than a metal	<input checked="" type="checkbox"/> anode
	2. one part of a voltaic cell, in which either oxidation or reduction occurs	<input checked="" type="checkbox"/> electrode
	3. a voltaic cell in which the electrolyte is a paste	<input checked="" type="checkbox"/> cathode
	4. the electrode at which reduction occurs	<input checked="" type="checkbox"/> dry cell
	5. a voltaic cell in which a fuel undergoes oxidation and from which electrical energy is obtained continuously	<input checked="" type="checkbox"/> battery
	6. an electrochemical cell that is used to convert chemical energy into electrical energy	<input checked="" type="checkbox"/> electrolysis
	7. a tube containing a conducting solution	<input checked="" type="checkbox"/> voltaic cell
	8. a group of cells that are connected together	<input checked="" type="checkbox"/> half-cell
	9. the process in which electrical energy is used to bring about a chemical change	<input checked="" type="checkbox"/> salt bridge
	10. the electrode at which oxidation occurs	<input checked="" type="checkbox"/> fuel cell

B. Multiple Choice

Choose the best answer and write its letter on the line.

11. In the electrolysis of brine, the substance produced at the cathode is
a. chlorine. c. sodium.
b. oxygen. d. hydrogen.
12. Which of the following describes a dry cell?
a. It can be recharged many times.
b. The graphite rod does not undergo reduction, even though it is the cathode.
c. It contains concentrated sulfuric acid.
d. all of the above.

High School Electrochemistry Study Guide Answers

R Barnett



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