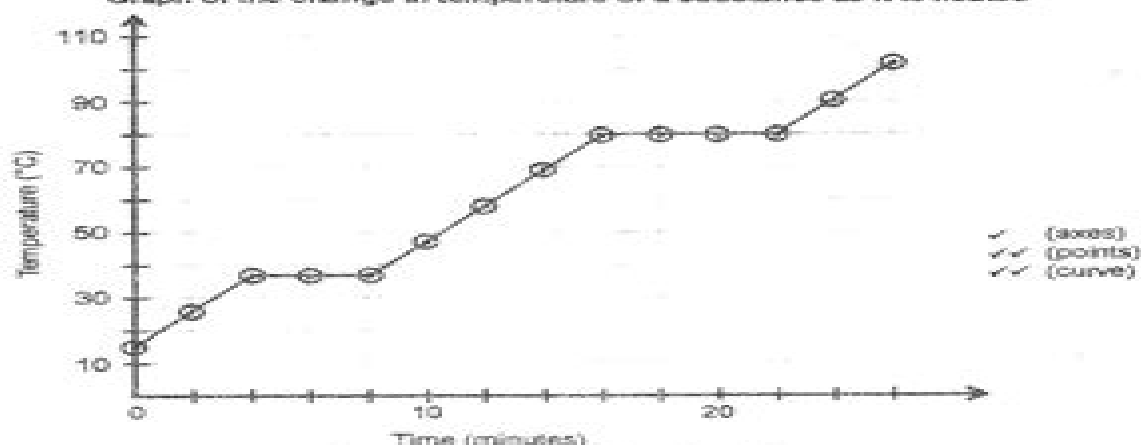


3.4

Graph of the change in temperature of a substance as it is heated

(5)



3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)

3.6 37 °C ✓ (2)

3.7 80 °C ✓ (1)

3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)

3.9 Physical ✓ (1)

3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

4.1

4.1.1 Homogeneous ✓ (1)

4.1.2 Homogeneous ✓ (1)

4.1.3 Pure substance ✓ (1)

4.2

4.2.1 Potassium bromide ✓ (2)

4.2.2 Calcium carbonate ✓ (2)

4.2.3 Ammonium chloride ✓ (2)

4.3

4.3.1 ZnSO_4 ✓✓ (2)4.3.2 KMnO_4 ✓✓ (2)4.3.3 H_2O ✓✓ (2)

[17]

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1. Understanding the eBook Grade Life Science Question Paper 2014
 - The Rise of Digital Reading Grade Life Science Question Paper 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Life Science Question Paper 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Life Science Question Paper 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Life Science Question Paper 2014
 - Personalized Recommendations
 - Grade Life Science Question Paper 2014 User Reviews and Ratings
 - Grade Life Science Question Paper 2014 and Bestseller Lists
5. Accessing Grade Life Science Question Paper 2014 Free and Paid eBooks
 - Grade Life Science Question Paper 2014 Public Domain eBooks
 - Grade Life Science Question Paper 2014 eBook Subscription Services
 - Grade Life Science Question Paper 2014 Budget-Friendly Options
6. Navigating Grade Life Science Question Paper 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade Life Science Question Paper 2014 Compatibility with Devices
 - Grade Life Science Question Paper 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Life Science Question Paper 2014
 - Highlighting and Note-Taking Grade Life Science Question Paper 2014
 - Interactive Elements Grade Life Science Question Paper 2014
8. Staying Engaged with Grade Life Science Question Paper 2014

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade Life Science Question Paper 2014
- 9. Balancing eBooks and Physical Books Grade Life Science Question Paper 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Life Science Question Paper 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade Life Science Question Paper 2014
 - Setting Reading Goals Grade Life Science Question Paper 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade Life Science Question Paper 2014
 - Fact-Checking eBook Content of Grade Life Science Question Paper 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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