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Assignment #2 - Compound Names and Formulas Multi-valent ions only

	ng compounds, all of which have been named using the ic charge is given after the first element (metallic) in Roman
1. copper (II) oxide	11. manganese (III) oxide My 203
2. mercury (I) oxide +\0,20	12. vanadium (II) bromide VB(z
3. gold (III) chloride _AuCl3	13. niobium(V) oxide No205
4. thallium (III) bromide TISV3	14. titanium (III) oxide
5. bismuth (V) oxide	15. titanium (III) nitride TIN

- 7. uranium (VI) oxide <u>VVO3</u> 17. cobalt (II) phosphide <u>CO3P3</u>
- 8. protectinium (IV) bromide PaBY4 18. tin (II) oxide Sn0
- 9. cerium (III) oxide <u>Ce2()3</u> 19. thulium (II) bromide <u>Tw/Br/2</u>
- 10. arsenic (V) sulphide A52SS 20. copper (II) bromide CUBYZ
- B. Name the compound using the Roman Numeral Method and balance the ionic charges.
- 1. SnCL Tin (IV) Chloride 9. Pdf. Palladium (IV) Fluoride
 2. BiBrs Bismuth (V) Bromide 10. OszO, Osmium (III) Oxide
- 3. POO2 POLONIUM (IV) OXICE 11. MOBEZ MOLYBOLENIUM (II) Bramide
- 4. Pbl. Lead (I) Iodide 12. Vol. Vanadium (V) Chloride
- 5. HgO <u>Hercury (II) Oxide</u> 13. Mn₂O₃ Manganese (III) Oxide 6. HgCl <u>Hercury (II) Chroride</u> 14. Coo <u>Colony (II) Oxide</u>
- 7. Augos Gold (III) Oxide 15. Np2Os NEptunium (III) Oxide
- 8. FeCI: Iron (II) Chloride 16. V2Os Vanadium (V) Oxide

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