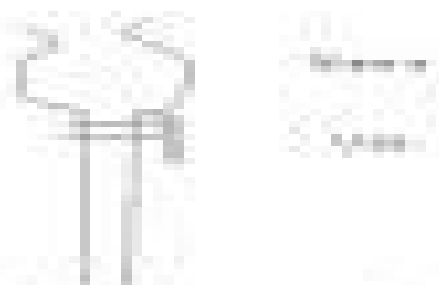


Figure 1: Chemistry: Electrode Potentials and Diagrams

Question 1:

(a) The following are standard conventional half-cells. Label them accordingly (10%)



(b) What are measures to be taken to avoid any errors in the measurement of pH? (10%)

Answer 1:

There is a standard solution of H+ ions, which is used to calibrate the pH meter.

(a) Standard solution of H+ ions is used to calibrate the pH meter. It is used to adjust the pH meter.

(b) There is a reason for the existence of pH meter.

(c) Standard solution of H+ ions is used to calibrate the pH meter. It is used to adjust the pH meter.

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O García

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